# Anticancer and Antifertility Agents. II. $N, N^{\prime}$-Hexamethylenebis(3-pyridinecarboxamide) 

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## Abstract

The centrosymmetric molecule $N, N^{\prime}$-hexamethylenebis-(3-pyridinecarboxamide), $\mathrm{C}_{18} \mathrm{H}_{22} \mathrm{~N}_{4} \mathrm{O}_{2}$, contains a crystallographic centre of inversion, with only half of the molecule constituting the asymmetric unit. There are intermolecular $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}=\mathrm{C}$ hydrogen bonds $[\mathrm{N} \cdots \mathrm{O}$ 2.996 (2) $\AA$ A present in the structure.

## Comment

Hexamethylenebisacetamide (HMBA) is an agent currently undergoing clinical trials that induces differentiation of certain types of cancer cells to non-malignant phenotypes (Andreeff et al., 1992). In order to discover more effective anticancer agents, identify structureactivity relationships, understand molecular mechanisms and provide structural data for drug design, a series of compounds structurally related to HMBA have been synthesized and their crystal structures determined (Zhang, Liu, Wei \& Shao, 1996). The title compound, (I), is one of these compounds.

(I)

The title compound (Fig. 1) can inhibit proliferation of lung cancer cells (Ao, Tanaka, Nakao, Yamagami \& Fujii, 1991) and it has been proven from pharmacological experiments (Zhang \& Wang, 1980; Zhang, Xing \& Wang, 1982) to possess antifertility activity in male rats. It can inhibit spermatogenesis over a long period and does not produce any toxic effects with effective dosage (Li, Qin, Zhang \& Li, 1979). Because of this, it may turn out to be an improved male sterilization drug.

Because intermolecular $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}=\mathrm{C}$ hydrogen bonds occur in HMBA (Bailey, 1955), this kind of intermolecular interaction could be assumed to occur in the title compound and is realised along the approximate direction of the $b$ axis, where there are intermolecular N-


Fig. 1. View of $\mathrm{C}_{18} \mathrm{H}_{22} \mathrm{~N}_{4} \mathrm{O}_{2}$ showing the labelling of the non- H atoms. Displacement ellipsoids are shown at $50 \%$ probability levels and H atoms are drawn as small circles of arbitrary radii.
$\mathrm{H} \cdots \mathrm{O}=\mathrm{C}$ hydrogen bonds of length 2.996 (2) $\mathrm{A}[\mathrm{O}=\mathrm{C}$ at $x, y-1, z]$.

## Experimental

The title compound was prepared by condensation of nicotinic acid and 1,6 -hexanediamine in a $2: 1$ molar ratio using Schlenk techniques. Recrystallization was from ethanol solution. The density $D_{m}$ was measured by flotation in $\mathrm{CHCl}_{3} / \mathrm{Cl}\left(\mathrm{CH}_{2}\right)_{3} \mathrm{Cl}$ solution.

## Crystal data

$\mathrm{C}_{18} \mathrm{H}_{22} \mathrm{~N}_{4} \mathrm{O}_{2}$
$M_{r}=326.4$
Monoclinic
$P 2_{1} / n$
$a=5.2240(10) \AA$
$b=5.0960(10) \AA$
$c=31.006$ (6) $\AA$
$\beta=94.53$ (3) ${ }^{\circ}$
$V=822.8$ (3) $\AA^{3}$
$Z=2$
$D_{x}=1.317 \mathrm{Mg} \mathrm{m}^{-3}$
$D_{m}=1.3167(10) \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
$\lambda=0.71073 \AA$
Cell parameters from 25
reflections
$\theta=6-10^{\circ}$
$\mu=0.083 \mathrm{~mm}^{-1}$
$T=296.0$ (1) K
Rod
$0.40 \times 0.20 \times 0.20 \mathrm{~mm}$
Colourless

## Data collection

Rigaku AFC-6S diffractometer
$2 \theta$ scans
Absorption correction: $\psi$ scans (Coppens, Leiserowitz \& Rabinovich, 1965)
$T_{\text {min }}=0.995, T_{\text {max }}=$ 1.000

2293 measured reflections 1892 independent reflections

## 1537 observed reflections

$[F \geq 4.0 \sigma(F)]$
$R_{\mathrm{mt}}=0.032$
$\theta_{\text {max }}=27.5^{\circ}$
$h=0 \rightarrow 6$
$k=0 \rightarrow 6$
$l=-39 \rightarrow 40$
3 standard reflections monitored every 150 reflections intensity decay: 3\%

## Refinement

Refinement on $F$
$(\Delta / \sigma)_{\text {max }}=0.003$
$R=0.0411$
$w R=0.0639$
$S=1.62$
$\Delta \rho_{\text {max }}=0.22 \mathrm{e}^{-3}$
$\Delta \rho_{\text {min }}=-0.16 \mathrm{e}^{-3}$
Extinction correction: none

1537 reflections
153 parameters
All H-atom parameters refined
$w=1 /\left[\sigma^{2}(F)+0.001 F^{2}\right]$

Atomic scattering factors from International Tables for X-ray Crystallography (1974, Vol. IV, Table 2.3.1)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters $\left(\AA^{2}\right)$

| $U_{\mathrm{eq}}=(1 / 3) \sum_{i} \sum_{j} U_{i j} a_{i}^{*} a_{j}^{*} \mathbf{a}_{i} \cdot \mathbf{a}_{j}$. |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | $x$ | $y$ | $z$ | $U_{\text {eq }}$ |
| 0 | 0.3725 (2) | 0.6828 (2) | 0.1022 (1) | 0.051 (1) |
| $N(1)$ | 0.4777 (2) | 0.2584 (2) | 0.0950 (1) | 0.039 (1) |
| N(2) | -0.1524 (2) | 0.5030 (3) | 0.1916 (1) | 0.053 (1) |
| C(1) | 0.6379 (3) | 0.3119 (2) | 0.0595 (1) | 0.040 (1) |
| C(2) | 0.7755 (3) | 0.0686 (3) | 0.0460 (1) | 0.039 (1) |
| C(3) | 0.9338 (3) | 0.1218 (3) | 0.0077 (1) | 0.038 (1) |
| C(4) | 0.3571 (2) | 0.4528 (2) | 0.1139 (1) | 0.036 (1) |
| C(5) | 0.2004 (2) | 0.3792 (2) | 0.1505 (1) | 0.036 (1) |
| C(6) | 0.2505 (3) | 0.1635 (3) | 0.1768 (1) | 0.043 (1) |
| C(7) | 0.0975 (3) | 0.1217 (3) | 0.2108 (1) | 0.050 (1) |
| C(8) | -0.0978 (3) | 0.2956 (3) | 0.2168 (1) | 0.052 (1) |
| C (9) | -0.0035 (3) | 0.5418 (3) | 0.1592 (1) | 0.043 (1) |

Table 2. Selected geometric parameters $\left(\AA{ }^{\circ},^{\circ}\right)$

| $\mathrm{O}-\mathrm{C}(4)$ | $1.232(2)$ | $\mathrm{C}(3)-\mathrm{C}\left(3^{\mathrm{i}}\right)$ | $1.517(3)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{N}(1)-\mathrm{C}(1)$ | $1.460(2)$ | $\mathrm{C}(4)-\mathrm{C}(5)$ | $1.498(2)$ |
| $\mathrm{N}(1)-\mathrm{C}(4)$ | $1.334(2)$ | $\mathrm{C}(5)-\mathrm{C}(6)$ | $1.381(2)$ |
| $\mathrm{N}(2)-\mathrm{C}(8)$ | $1.331(2)$ | $\mathrm{C}(5)-\mathrm{C}(9)$ | $1.392(2)$ |
| $\mathrm{N}(2)-\mathrm{C}(9)$ | $1.334(2)$ | $\mathrm{C}(6)-\mathrm{C}(7)$ | $1.386(2)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)$ | $1.509(2)$ | $\mathrm{C}(7)-\mathrm{C}(8)$ | $1.375(2)$ |
| $\mathrm{C}(2)-\mathrm{C}(3)$ | $1.523(2)$ |  |  |
| $\mathrm{C}(1)-\mathrm{N}(1)-\mathrm{C}(4)$ | $120.7(1)$ | $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | $123.9(1)$ |
| $\mathrm{C}(8)-\mathrm{N}(2)-\mathrm{C}(9)$ | $116.6(1)$ | $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(9)$ | $118.0(1)$ |
| $\mathrm{N}(1)-\mathrm{C}(1)-\mathrm{C}(2)$ | $111.7(1)$ | $\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{C}(9)$ | $118.1(1)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | $111.7(1)$ | $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(7)$ | $118.5(1)$ |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}\left(3^{1}\right)$ | $112.9(1)$ | $\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(8)$ | $118.9(1)$ |
| $\mathrm{O}-\mathrm{C}(4)-\mathrm{N}(1)$ | $122.2(1)$ | $\mathrm{N}(2)-\mathrm{C}(8)-\mathrm{C}(7)$ | $124.0(1)$ |
| $\mathrm{O}-\mathrm{C}(4)-\mathrm{C}(5)$ | $120.9(1)$ | $\mathrm{N}(2)-\mathrm{C}(9)-\mathrm{C}(5)$ | $123.9(1)$ |
| $\mathrm{N}(1)-\mathrm{C}(4)-\mathrm{C}(5)$ | $116.9(1)$ |  |  |

Symmetry code: (i) $2-x,-y,-z$.
The structure was solved and refined using SHELXTL-Plus (Sheldrick, 1990). All calculations were carried out using SHELXTL-Plus and the figure was produced with PLATON (Spek, 1990).

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Lists of structure factors, anisotropic displacement parameters, H atom coordinates, complete geometry and torsion angles have been deposited with the IUCr (Reference: TA1081). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CHI 2HU, England.

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## $17 \alpha$-Ethynyl-19-nor-3-oxo-4-androsten$17 \beta$-yl Acetate (Norethindrone Acetate)

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## Abstract

The steroid skeleton of the title compound, $\mathrm{C}_{22} \mathrm{H}_{28} \mathrm{O}_{3}$, is almost identical to the parent molecule norethindrone. The ethynyl residue donates a $\mathrm{C} \equiv \mathrm{C}-\mathrm{H} \cdots \mathrm{O}=\mathrm{C}$ hydrogen-bond interaction, with an $\mathrm{H} \cdots \mathrm{O}$ separation of $2.18 \AA$.

## Comment

$17 \alpha$-Ethynyl steroids are widely used as constituents of contraceptive agents and numerous crystal structures have been determined (e.g. van Geerestein, 1988). The structure of the parent molecule of the title compound, norethindrone, was reported by Mornon, Lepicard \& Delettre (1976). Recently, Lutz, van der Maas \& Kanters (1994) pointed out that ethynyl steroids may serve as a model to study $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ interactions donated by the ethynyl residue $\mathrm{C} \equiv \mathrm{C}-\mathrm{H}$. The latter is known to be one of the most acidic $\mathrm{C}-\mathrm{H}$ groups and therefore one of the strongest $\mathrm{C}-\mathrm{H}$ hydrogen-bond donors (see Desiraju, 1991; Steiner, 1994a). In this context, the crystal structure of the title compound (1) was determined.

(1)

The molecular conformation of the steroid skeleton (Fig. 1) is almost identical with that of norethindrone (Mornon, Lepicard \& Delettre, 1976), and therefore

